

# Notes for Lectures on Quantum Mechanics \*

## Particle in a Rigid Spherical Box

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The potential for a rigid spherical box can be written as

$$V(r) = \begin{cases} 0, & 0 < r < a \\ \infty, & r > a \end{cases}. \quad (1)$$

The problem is separable in spherical polar coordinates and form of the full wave function is

$$\psi(r, \theta, \phi) = R(r)Y_{\ell m}(\theta, \phi). \quad (2)$$

We need to consider solutions of the radial equation only. No solution can be found for  $E < 0$ , therefore we consider  $E > 0$ . For  $r > a$  the potential is infinite and hence the radial wave function must be zero, Next we consider  $r < a$ , where the potential is zero. The radial equation assumes the form

$$-\frac{1}{r^2} \frac{d}{dr} r^2 \frac{dR(r)}{dr} + \frac{\ell(\ell+1)\hbar^2}{2mr^2} R(r) - ER(r) = 0. \quad (3)$$

The most general solution of this equation is given in terms of spherical Bessel functions  $j_{\ell}, n_{\ell}$  and we write it as

$$R_{E\ell}(r) = Aj_{\ell}(kr) + Bn_{\ell}(kr), \quad k^2 = \frac{2mE}{\hbar^2}. \quad (4)$$

Recall that near  $r = 0$ ,  $n_{\ell}(r) \sim r^{-\ell-1}$  and blows up as  $r \rightarrow 0$ . Therefore we must set  $B = 0$  if the solution is to remain finite at  $r = 0$ . Thus we get

$$R_{\ell}(r) = \begin{cases} Aj_{\ell}(kr), & 0 < r < a \\ 0 & r > a \end{cases}. \quad (5)$$

Next we must demand that the radial wave function  $R(r)$  must be continuous at  $r = a$ . Remember that there is no corresponding requirement on the derivative for this case of infinite jump in the potential at  $r = a$  The continuity requirement of  $R_{E\ell}(r)$  becomes

$$j_{\ell}(ka) = 0. \quad (6)$$

The solutions of the above equation determine allowed values of  $k$  and hence allowed bound state energies.

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## Energy levels and degeneracy

To get all the solutions, one proceeds as follows. First set  $\ell = 0$  and locate the roots of  $j_0(ka) = 0$ . We call the roots as  $\rho_{0n}, n = 0, 1, 2, \dots$  and the corresponding energies are given by  $E = \frac{\hbar^2 \rho_{0n}^2}{2ma^2}$ . Here  $n$  denotes the number of nodes of the radial wave function for  $\ell = 0$ .

Next, we set  $\ell = 1$  and find the roots of  $j_1(kr) = 0$ , calling these roots as  $\rho_{1n}, n = 0, 1, 2, \dots$  the  $\ell = 1$  energy levels are given by  $E = \frac{\hbar^2 \rho_{1n}^2}{2ma^2}$ . This process is to be repeated for all values of angular momentum  $\ell$  and the number of bound states for each  $\ell$  turns out to be infinite. The states of definite energy depend on quantum numbers  $n\ell m$  and the energy does not depend on magnetic quantum number  $m$ . Therefore for a given azimuthal quantum number  $\ell$  we have  $(2\ell + 1)$  wave functions  $N_{n\ell} R_{n\ell}(r/\rho_{n\ell}) Y_{\ell m}(\theta, \phi), (m = -\ell, -\ell + 1, \dots, \ell)$  and the energy levels  $E_{n\ell}$  are  $(2\ell + 1)$  fold degenerate. The energy increases with  $\ell$  and also with increasing  $n$ . Thus schematic energy level diagram would appear as follows.

$n' = 3$	<u><math>n' = 3</math></u>	<u><math>n' = 3</math></u>
$n' = 2$	<u><math>n' = 2</math></u>	<u><math>n' = 2</math></u>
$n' = 1$	<u><math>n' = 1</math></u>	<u><math>n' = 1</math></u>
$n' = 0$	<u><math>n' = 0</math></u>	<u><math>n' = 0</math></u>
$l = 0$	$l = 1$	$l = 2$
nondegenerate	$m = -1, 0, 1$	$m = -2, -1, 0, 1, 2$
	3 fold degenerate	5 fold degenerate